

1. CO carbon monoxide
2. NaBr sodium bromide
3. P₂O₅ diphosphorous pentaoxide
4. Ti(SO₄)₂ titanium (IV) sulfate
5. FePO₄ iron (III) phosphate
6. CO₂ carbon dioxide
7. CF₄ carbon tetrafluoride
8. K₃N potassium nitride
9. SO₂ sulfur dioxide
10. CuOH copper (I) hydroxide

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| 11. | Zn(NO ₂) ₂ | zinc nitrate |
| 12. | VCl ₄ | vanadium (IV) chloride |
| 13. | N ₂ O | dinitrogen monoxide |
| 14. | N ₂ O ₄ | dinitrogen tetraoxide |
| 15. | BCl ₃ | boron trichloride |
| 16. | Mg ₃ P ₂ | magnesium phosphide |
| 17. | Sr(HCO ₃) ₂ | strontium bicarbonate |
| 18. | C ₂ S ₄ | dicarbon tetrasulfide |
| 19. | Ba ₃ N ₂ | barium nitride |
| 20. | NH ₃ | ammonia |

Write the formula for each of the following compounds:

1. silicon dioxide SiO_2
2. nickel (III) sulfide Ni_2S_3
3. copper (II) sulfate CuSO_4
4. manganese (II) phosphate $\text{Mn}_3(\text{PO}_4)_2$
5. diboron tetrabromide B_2Br_4
6. aluminum phosphate AlPO_4
7. magnesium hydroxide $\text{Mg}(\text{OH})_2$
8. potassium carbonate K_2CO_3
9. ammonium oxide $(\text{NH}_4)_2\text{O}$
10. tin (IV) selenide Sn_3Se_4

11. cobalt (II) selenide Co_3Se_3
12. carbon tetrachloride CCl_4
13. nitrogen monoxide NO
14. sulfur dioxide SO_2
15. hydrogen peroxide H_2O_2
16. lithium nitrate LiNO_3
17. potassium bicarbonate KHCO_3
18. dicarbon tetra sulfide C_2S_4
19. nitrogen tribromide NBr_3
20. copper (I) sulfide C_2S