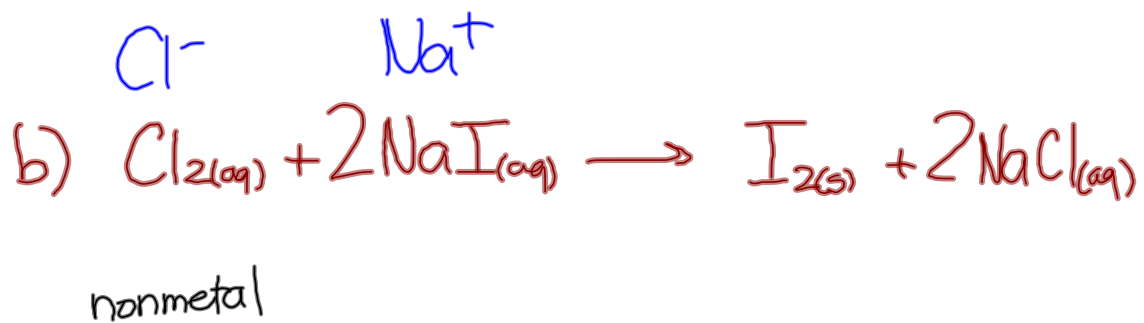
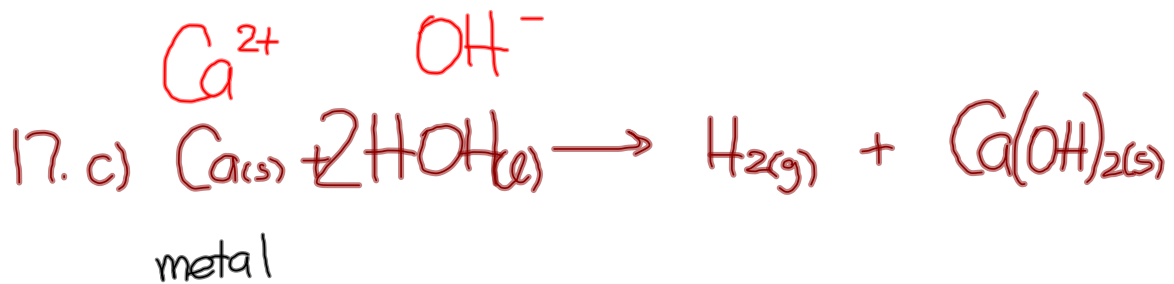


Homework - #17



Practice Problems

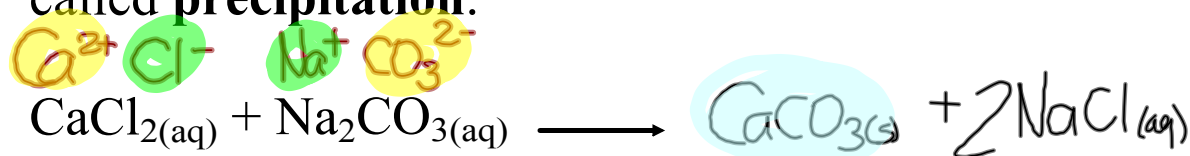


Chemical Reactions

V. Double Replacement Reaction

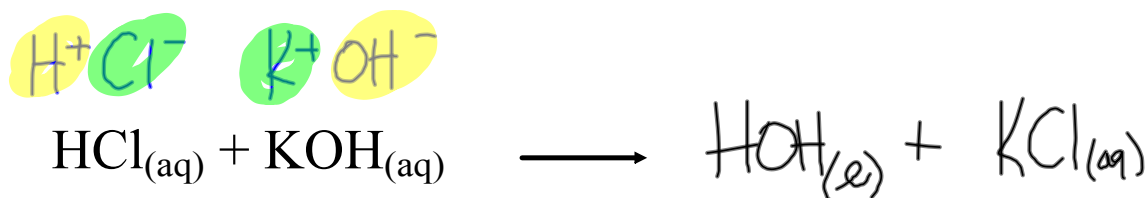
Reaction that occurs between two ionic compounds in solution. Ions will "change partners".

⇒ if one of the products has low solubility, it may form a precipitate (solid). This double replacement reaction is called **precipitation**.



compound + compound \longrightarrow **compound + compound**

A second type of double replacement reaction is a **neutralization** reaction, which is a reaction between an acid and a base, to form water and an ionic compound.



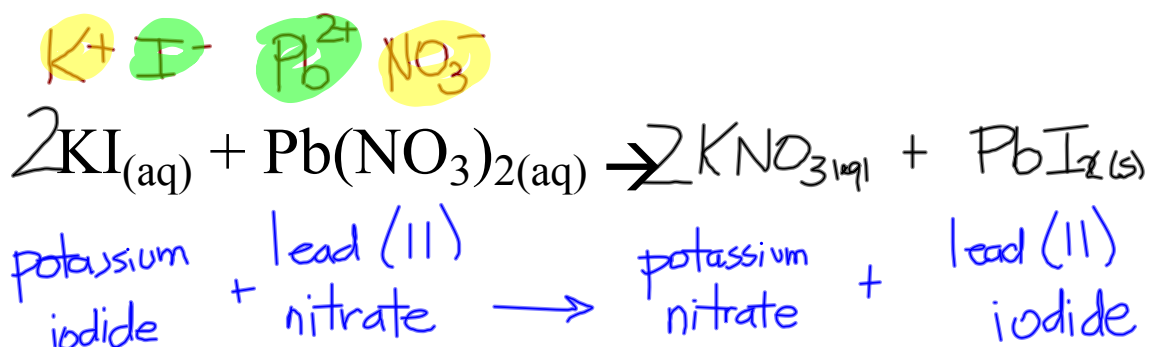
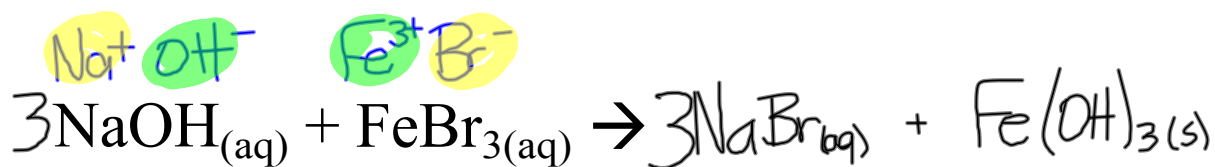
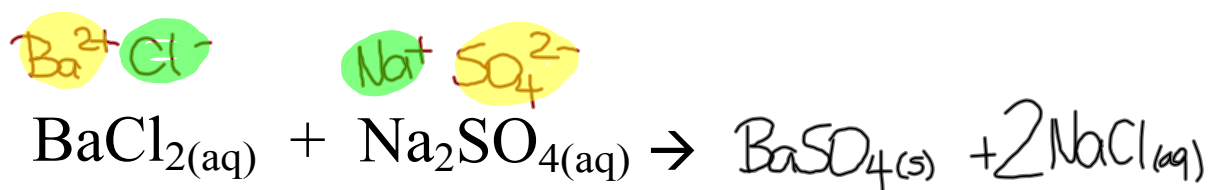
acid

base

water

**ionic
compound**

Practice Problems



Worksheet

p. 335 #18,19

p. 339 #22-27