Thursday Dec 15

Chapter 6: Chemical Reactions

Types of Chemical Equations:

- Word vs Chemical

Warm- Up

What chemical reactions do you conduct every morning?

Chemical Reactions

What is a chemical reaction?

- change that takes place when two or more substances (reactants) change into one or more new substances.
- the event resulting from the mixing of different substances
- a process in which one or more substances are changed into others

What is a chemical equation?

 a written representation of a chemical reaction using chemical formulas for reactant and products

What is a word equation?

 one way of representing a chemical equation using the names of reactants and products instead of chemical formulas

Chemical Equations

"reactants"
(starting materials)

"products" (finishing materials)

Word Equations

Examples:

Write the chemical equation to represent the following:

a) copper + silver nitrate
$$\Rightarrow$$
 silver + copper (II) nitrate
Cu + AgNO₃ \Rightarrow Ag + Cu(NO₃)₂

Examples:

Write the word equation to represent the following:

a)2
$$H_2 + O_2 \Rightarrow 2 H_2O$$

hydrogen + oxygen \Rightarrow water

b) Fe and O_2 react to form Fe_2O_3 iron + oxygen \Rightarrow iron (III) oxide

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