## Lab books due in ten minutes

## Check Worksheet

NaOH
$\mathrm{Na}^{+} \mathrm{OH}^{-}$

HCl
$\mathrm{H}^{+} \mathrm{Cl}^{-}$
(bb) a) magnesium hydroxide

$$
\begin{array}{ll}
\mathrm{Mg}^{2+} & \mathrm{OH}^{-} \quad \mathrm{Mg}(\mathrm{OH})_{2} \\
\mathrm{CH}^{-}
\end{array}
$$

b) hydrofluoric acid

$$
H^{+} F^{-}
$$

.ide

c) phosphoric acid

$$
\begin{aligned}
& \mathrm{H}^{+} \mathrm{PO}_{4}^{3-} \text {-at } \\
& \mathrm{H}^{+} \\
& \mathrm{H}^{+}
\end{aligned}
$$

d) lithium hydroxide

$$
\mathrm{Li}^{+} \mathrm{OH}^{-}
$$



Ina) KOH
$\mathrm{K}^{+} \mathrm{OH}^{-}$
b) HI
$H^{+}$I-
potassium hydroxide aqueous hydrogen iodide hydroiodic acid

$$
\begin{aligned}
& \mathrm{C} \mathrm{H}_{2} \mathrm{SO}_{4} \quad \text { aqueous hydrogen sulfate } \\
& 1+\mathrm{ch}^{2-}
\end{aligned}
$$

## Homework

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