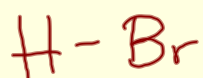


p. 239

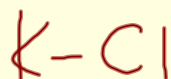
#30, 31

mod. polar



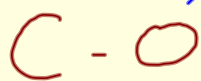
2.1 2.8

ionic



0.8 3.0

mod/very polar



2.5 3.5

mod/very polar



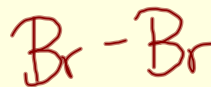
3.0 4.0

1.0 3.5



ionic

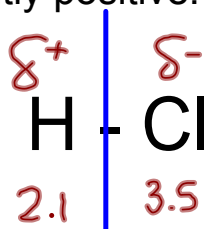
2.8 2.8



nonpolar

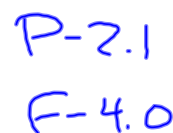
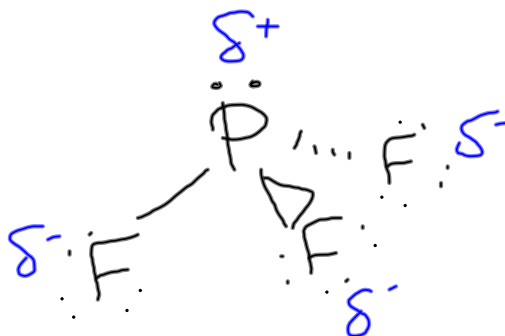
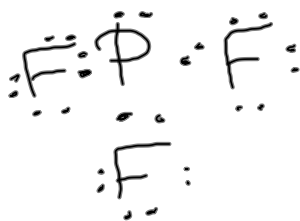
Polar Molecules

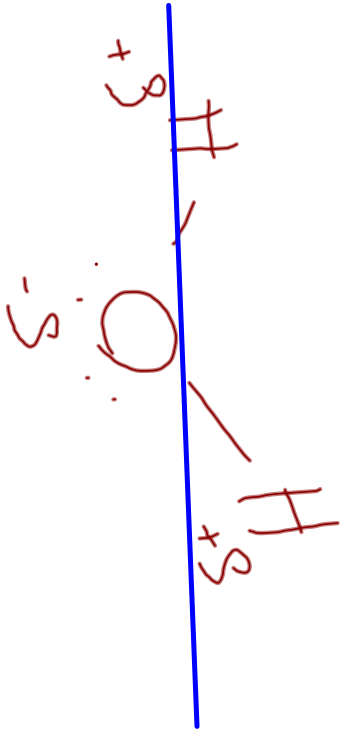
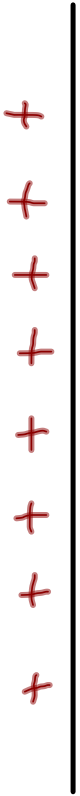
In a polar molecule, one end of the molecule is slightly negative, and the other end is slightly positive.

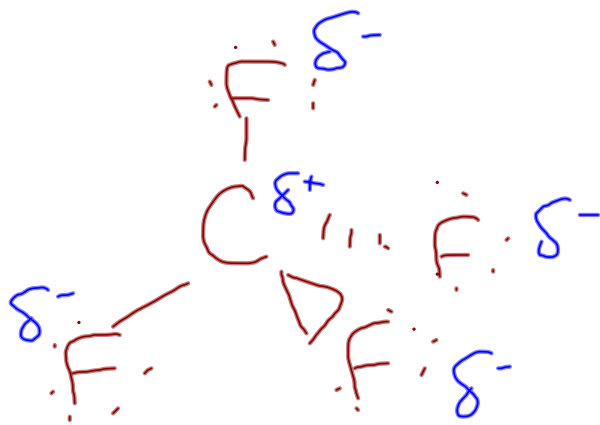


-Partial charges are often called charged regions or poles.

A molecule with two poles is called a **dipole**.







C → 2.5
F → 4.0

Review Worksheet
Hybridization and Polarity