Alcohols and Ethers Worksheet

c)
$$2$$
-hexanol + HCI — water + HCI + 1 -hexane + 2 -hexane

$$-\frac{1}{C}\cdot\frac{1}{C}\cdot\frac{1}{C}\cdot\frac{1}{C}\cdot\frac{1}{C}\cdot\frac{1}{C}\cdot\frac{1}{C}$$

H of H

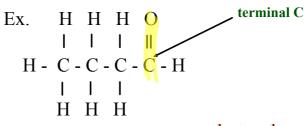
ADDITION

Aldehydes and Ketones

Aldehydes - contain a carbonyl group on a terminal carbon

- are named by replacing the "e" in alkane with al
- begin numbering at the end beginning with the aldehyde functional group

$$\begin{array}{c} O \\ \text{ } \\ \text{ } \\ \text{R(H) - C - H} \end{array}$$



butanal

Ketones - have a carbonyl on any carbon but the end carbon - are named by replacing "e" on the parent alkane with -one.

Aldehydes and ketones with the same number of carbons are isomers.

Organic Halides R-X

Alcohols R-OH

Ethers R-O-R'

Aldehydes R(H)-C-H

Ketones

2-pentanone

3-iodohexanal

3,3-dimothy/butanal

4-ethyl-3-methylhexanal

fluoropropanone

Aldehydes and Ketones Worksheet