

Nelson Chemistry Periodic Table of the Elements (2nd Edition)

2	He	4.00	—	—	0.179
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10	Ne	20.18	—	—	0.900
9	F	19.00	—	—	1.70
8	O	16.00	—	—	1.43

18	Ar	39.95	—	—	1.78
17	Cl	35.45	—	—	3.21
16	S	32.06	—	—	2.07

36	Kr	83.80	—	—	3.74
35	Br	79.90	—	—	3.12
54	Xe	131.29	—	—	5.89

86	Rn	222	—	—	9.73
85	At	(210)	—	—	—
118	Og	(294)	—	—	—

118	Og	(294)	—	—	—
117	Ts	(293)	—	—	—
116	Lv	(292)	—	—	—

1	H	1.01	1.01	—	—
2	He	4.00	—	—	0.179

9.01	Be	9.01	1.85	—	—
4	B	10.81	2.34	—	—
12	C	12.01	2.26	—	—

11	Li	6.94	0.534	—	—
12	Be	9.01	1.85	—	—
13	B	10.81	2.34	—	—

20	Ca	40.08	1.54	—	—
21	Sc	44.96	2.99	—	—
22	Ti	47.88	4.54	—	—

37	Rb	85.47	2.6	—	—
38	Sr	87.62	2.6	—	—
39	Y	88.91	4.47	—	—

55	Cs	132.91	1.88	—	—
56	Ba	137.33	3.5	—	—
57	La	138.91	6.15	—	—

87	Fr	(223)	—	—	—
88	Ra	(226)	—	—	—
104	Uuq	(261)	—	—	—

58	Ce	140.12	6.69	—	—
59	Pr	140.91	6.64	—	—
60	Nd	144.24	7.01	—	—

90	Th	232.04	11.7	—	—
91	Pa	231.04	15.4	—	—
92	U	238.03	19.1	—	—

26	Fe	55.85	7.87	—	—
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atomic number
 atomic molar mass (g/mol)
 melting point (°C)
 boiling point (°C)
 density of solids, liquids (g/cm³)
 density of gases (g/L)
 other ion charge
 common ion charge
 electronegativity
 other ion charge

element name
 synthetic in blue
 liquids in green
 gases in red

Values from
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 71st Edition

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