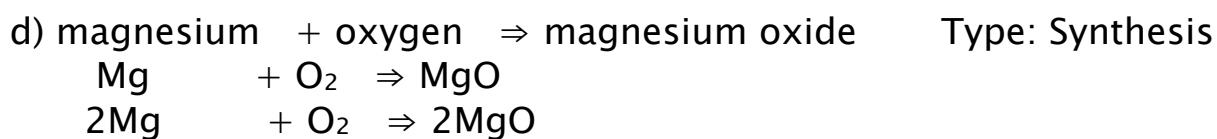
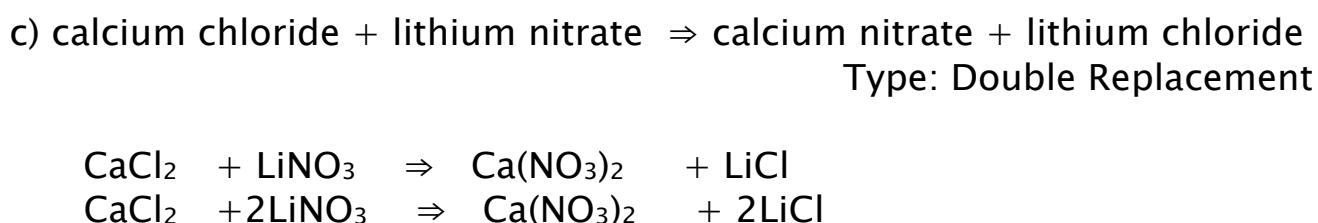
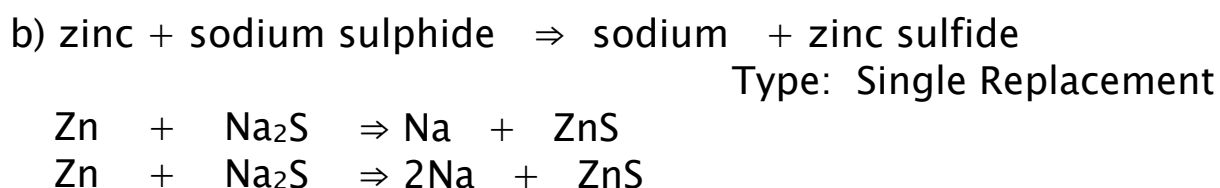
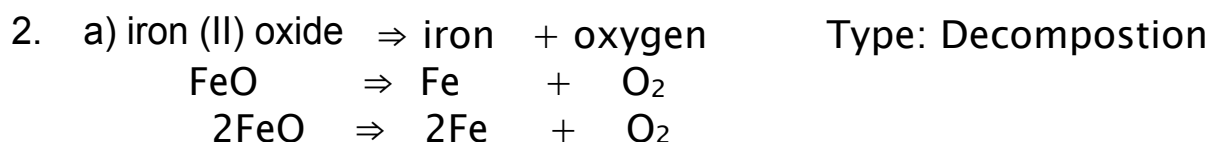
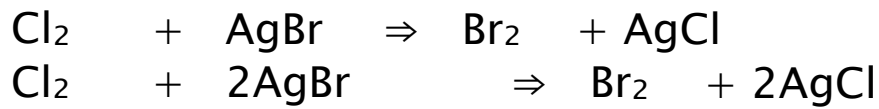


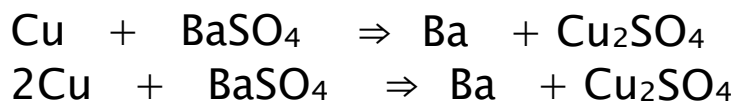
Answers
Physical Science 10
Review: Chemical Reactions



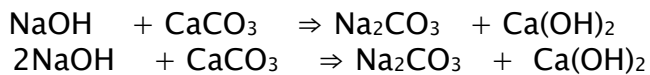
e) chlorine + silver bromide \Rightarrow bromine + silver chloride
Type: Single Replacement



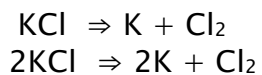
f) copper + barium sulphate \Rightarrow barium + copper (I) sulphate
Type: Single Replacement

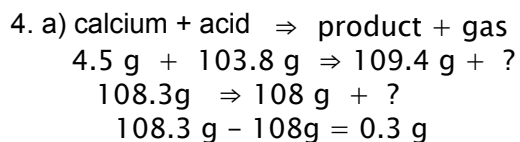
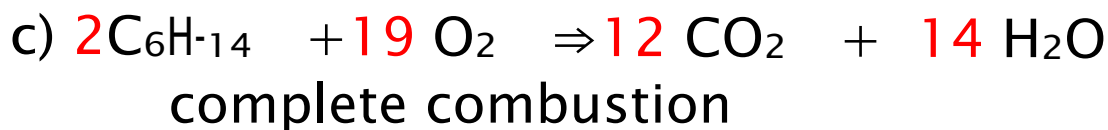
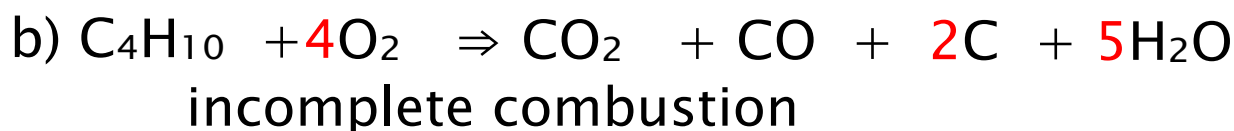
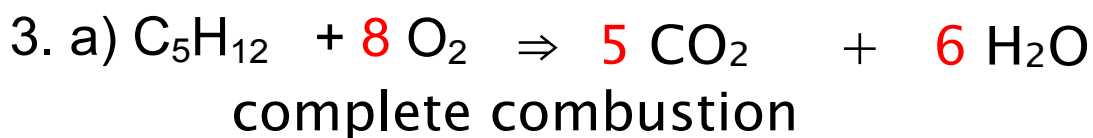


g) sodium hydroxide + calcium carbonate \Rightarrow sodium carbonate + calcium hydroxide
Type: Double Replacement



h) potassium chloride \Rightarrow potassium + chlorine
Type: Decomposition





b) The law of conservation of mass states that the mass of the reactants must equal the mass of the products in a reaction. The mass that is missing is the mass of the gas because it has evaporated into the air and cannot be measured in the beaker.