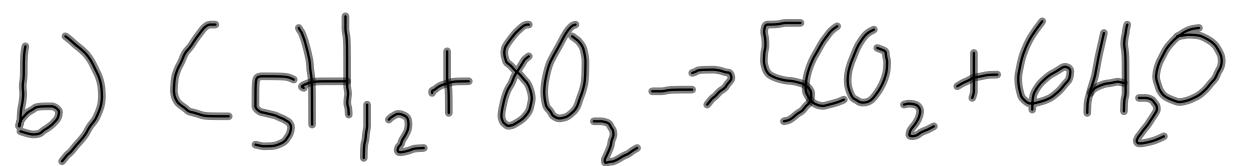
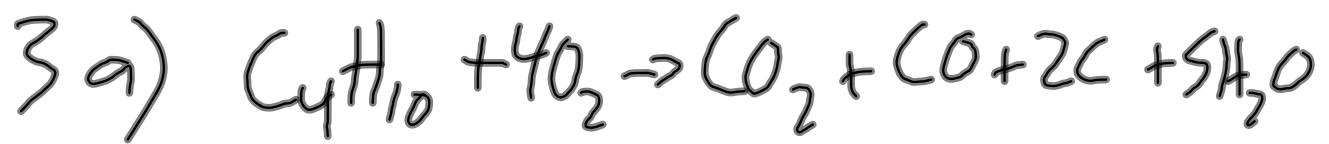


- a) $\text{Ca} + \text{Br}_2 \rightarrow \text{CaBr}_2$ synthesis
- b) $\text{Mg} + 2\text{AgCl} \rightarrow 2\text{Ag} + \text{MgCl}_2$ S.R.
- c) $2\text{NaF} + \text{ZnO} \rightarrow \text{Na}_2\text{O} + 2\text{ZnF}_2$ D.R.
- d) $\text{Fe}(\text{NO}_3)_2 + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + 2\text{CuNO}_3$
D.R.
- e) $2\text{ZnO} \rightarrow 2\text{Zn} + \text{O}_2$ decomposition



Warm-Up!

How many significant figures are in the following numbers:

a) 10.05 4

b) 0.0010100 5

c) 10,000 1

Addition and Subtraction



$$a) \underline{45} + 0.034 + 25.7 = \underline{70.734} = 71$$

$$b) \underline{123} - 45.6 = \underline{77.4} = 77$$

$$c) \underline{1000} + 456 - 12 = \underline{1444} = 1000$$



$$\begin{array}{r} \\ \underline{11} \\ 123.571 \end{array}$$

Multiplication and Division

- a) $35.152 \times \underline{1256} = 44,150.912 = 44,150$
- b) $456 / \underline{32} = 14.25 = 14$
- c) $\underline{34} \times 125 / 3 = 1416.\underline{6} = 1000$

120000

Ticket

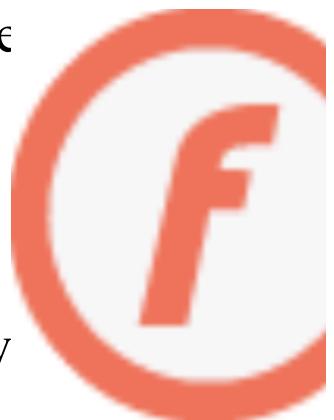
Calculate the number of significant digits in ϵ of the following:

- 1) 30803
- 2) 0.005670
- 3) 3450

Complete each of the following calculations giving the correct number of SF in your answer

$$4) 3.76 \text{ g} + 14.83 \text{ g} + 2.1 \text{ g} =$$

$$5) 34.8 / 2.6 =$$



Homework - Significant Figures Worksheet

Attachments

S10 sig figs answers.notebook