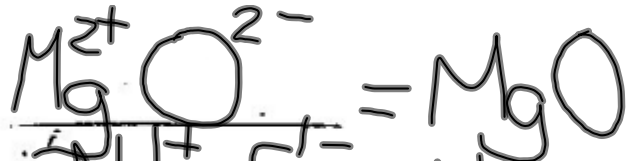
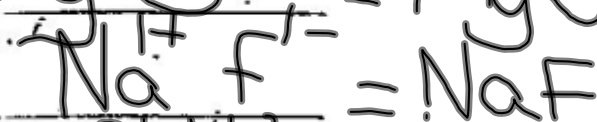


(a) magnesium oxide



(b) sodium fluoride



(c) aluminum nitride



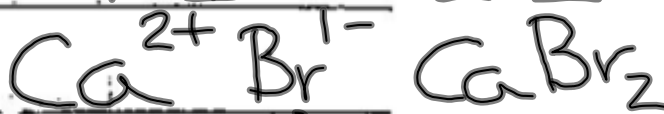
(d) potassium sulfide



(e) lithium iodide



(f) calcium bromide



(g) beryllium oxide



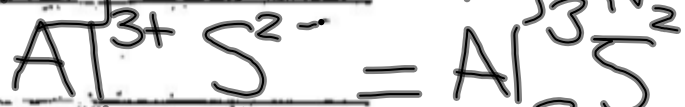
(h) nickel chloride



(i) magnesium nitride



(j) aluminum sulfide



(k) copper(I) bromide



(l) tin(II) iodide



(m) iron(III) chloride



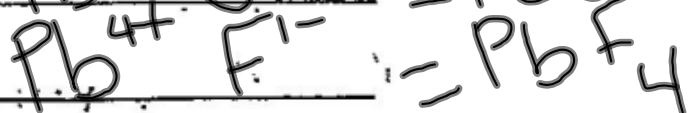
(n) calcium phosphide



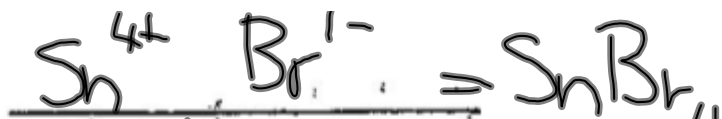
(o) lead(II) oxide



(p) lead(IV) fluoride



(q) tin(IV) bromide



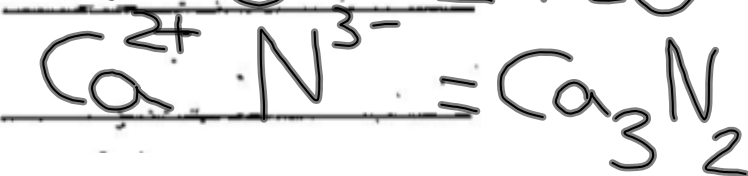
(r) copper(II) sulfide



(s) iron(II) oxide



(t) calcium nitride



2. Write the names for the following compounds.

(a)  $\text{Li}_2\text{O}$  lithium oxide

(b)  $\text{AlCl}_3$  aluminum chloride

(c)  $\text{MgS}$  magnesium sulfide

(d)  $\text{CaO}$  calcium oxide

(e)  $\text{KBr}$  potassium bromide

(f)  $\text{BeF}$  beryllium fluoride

(g)  $\text{Na}_3\text{N}$  sodium nitride

(h)  $\text{Al}_2\text{O}_3$  aluminum oxide

- (i)  $\text{CuCl}_2$  copper (II) chloride
- (j)  $\text{FeBr}_3$  iron (III) bromide
- (k)  $\text{PbS}$  lead (II) sulfide
- (l)  $\text{SnO}_2$  tin (IV) oxide
- (m)  $\text{Na}_2\text{S}$  sodium sulfide
- (n)  $\text{Mg}_3\text{P}_2$  magnesium phosphide
- (o)  $\text{NiO}$  nickel (II) oxide
- (p)  $\text{CuI}$  copper (I) iodide

- (q)  $\text{PbCl}_4$  lead (IV) chloride
- (r)  $\text{FeP}$  iron (III) phosphide
- (s)  $\text{CaF}_2$  calcium fluoride
- (t)  $\text{K}_3\text{P}$  potassium phosphide

## Attachments

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answers pg 198 #1-7.notebook