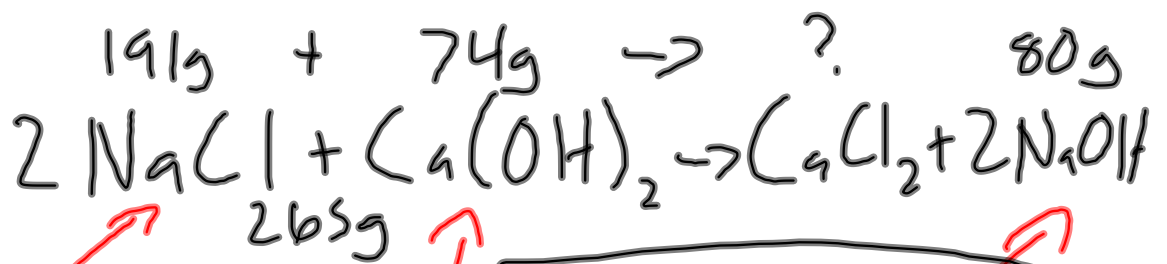




I failed you :)

lab: Wed.
Test Tues.
↑
Tomorrow



$$\text{NaCl} = 191\text{g}$$

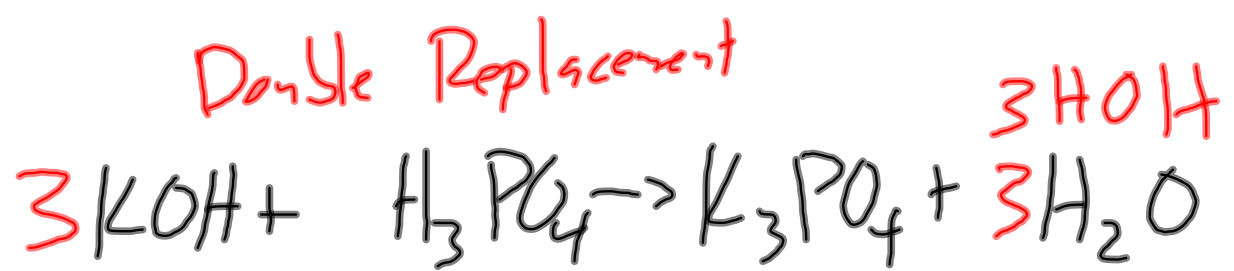
$$\text{Ca(OH)}_2 = 74\text{g}$$

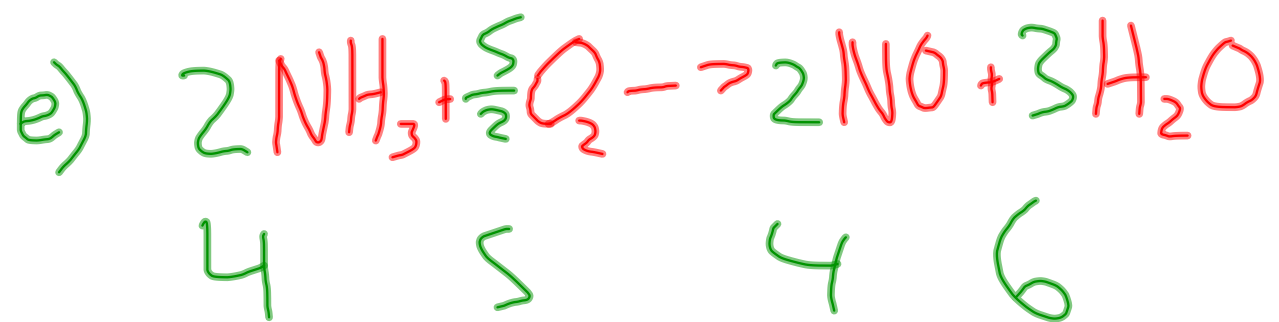
$$\text{NaOH} = 80\text{g}$$

$$\text{CaCl}_2 = ?\text{g} = 145\text{g}$$

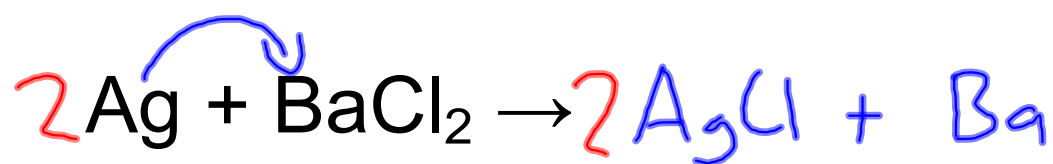
Law of
Conservation of
mass

Double Replacement



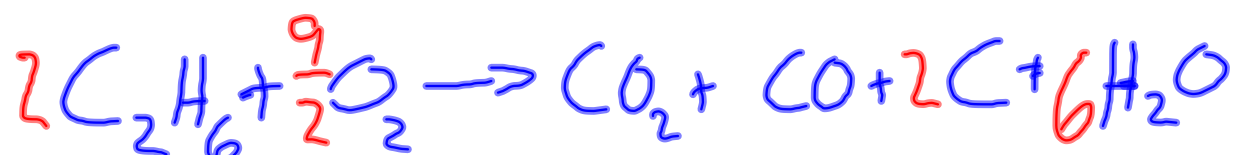


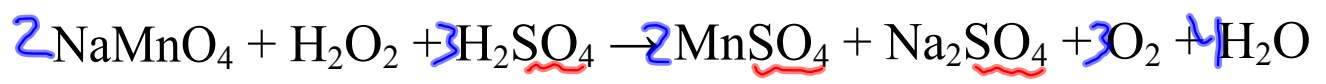
The following is a single replacement reaction:



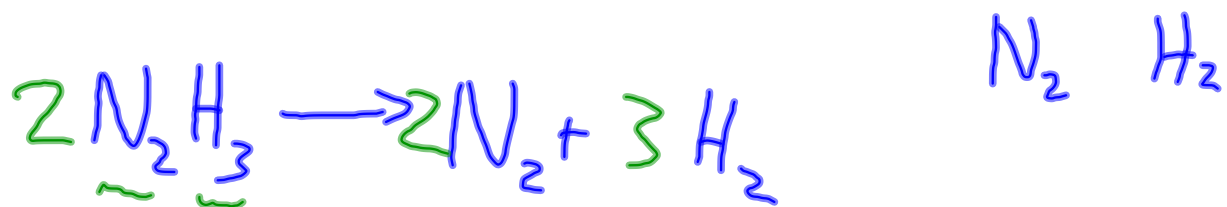
predict the products and balance the equation

dicarbon hexahydride experiences incomplete combustion. Write the equation and balance it.



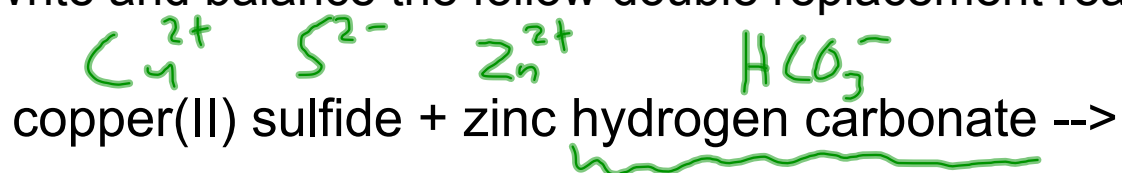


a) Write the equation for when dinitrogen trihydride decomposes



b) Balance it

Write and balance the follow double replacement reaction:



Write and balance the following equation. What type of equation is it?

