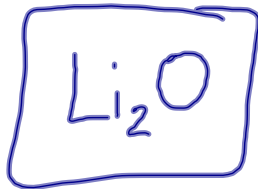


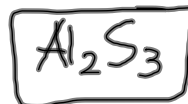
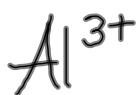
# BINARY IONIC COMPOUNDS TYPE I

lithium and oxygen

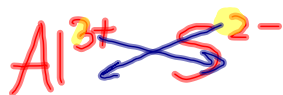
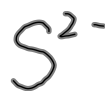


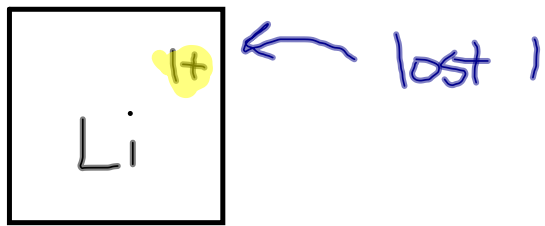
lithium oxide

aluminum and sulfur



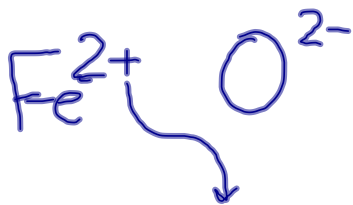
aluminum sulfide



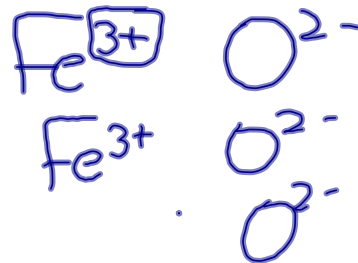


## BINARY IONIC COMPOUNDS TYPE II

Write name for  $\text{FeO}$  :

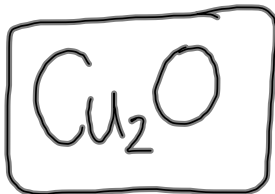


iron(II) oxide



iron(III) oxide

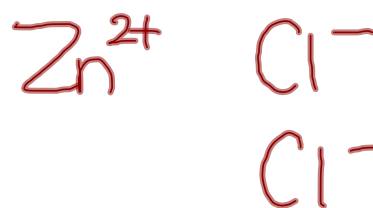
Copper (I) oxide



**Write the chemical formula of the following ionic compounds:**

**(a) iron (II) fluoride**

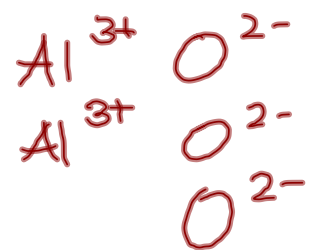
**(b) zinc chloride**



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Write the name of the following ionic compounds:

(a)  $\text{Al}_2\text{O}_3$



aluminum oxide

(b)  $\text{PbO}$



lead (II) oxide