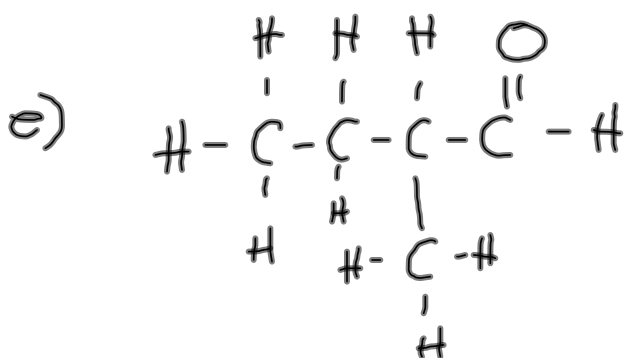
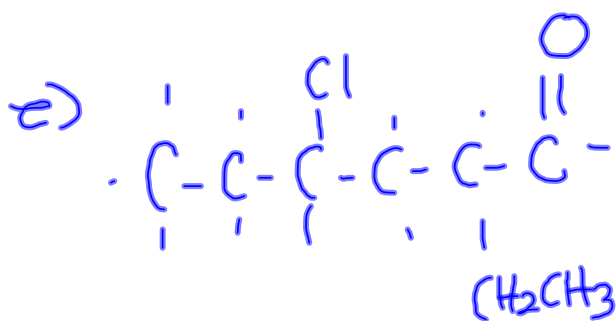
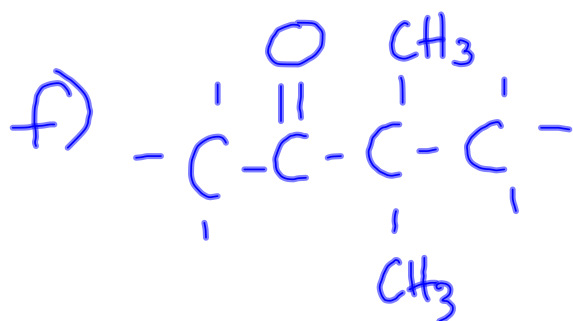
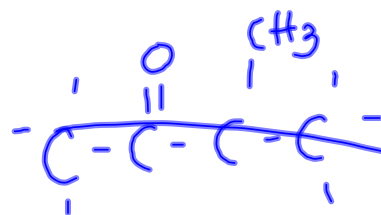
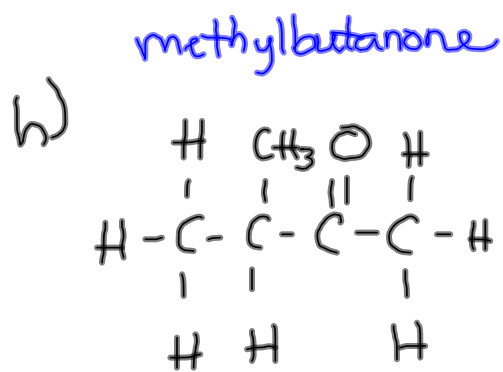
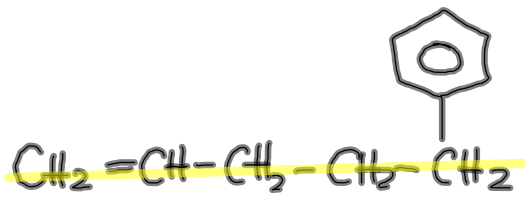


# Ketones and Aldehydes Worksheet

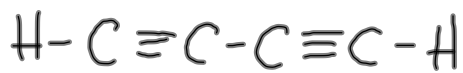


2-methylbutanal





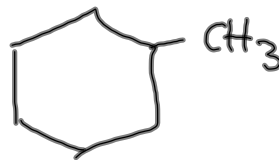
5-phenyl-1-pentene



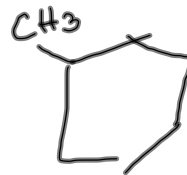
butadiyne



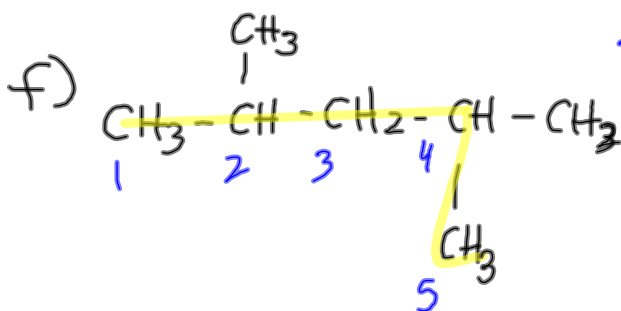
methylcyclopentane

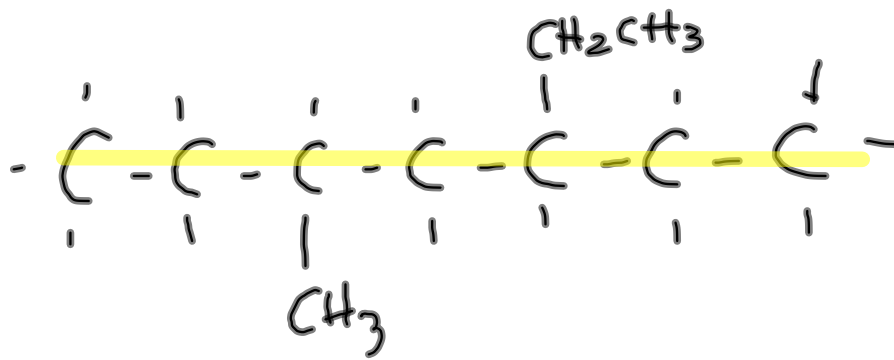


methylcyclohexane



2,4-dimethylhexane

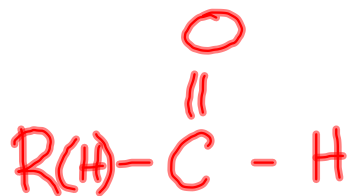




Organic Halides



Aldehydes



Ketones



Alcohols



Ethers



# Alcohols

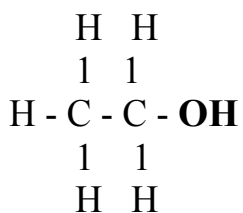
## Alcohols

- hydrocarbon derivatives containing a hydroxide (**OH**) functional group

## Naming

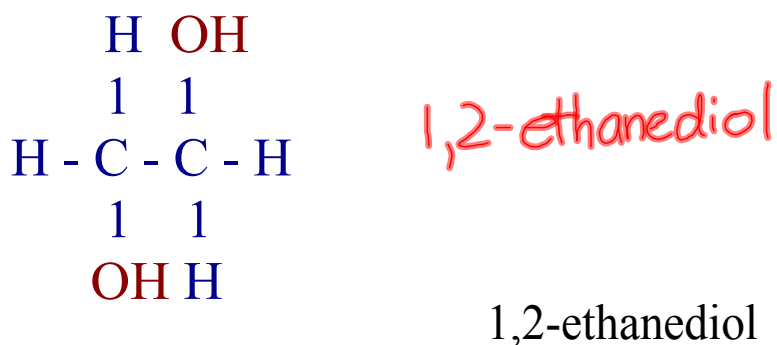
When naming alcohols, the -e is dropped from the name of the simple alkane, and it is replaced by an **-ol**.

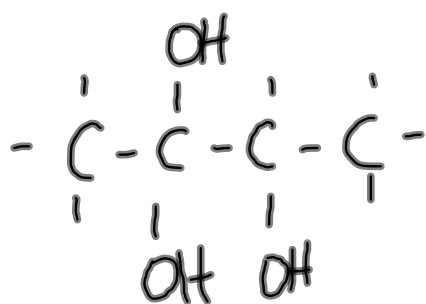
Ex. ethanol



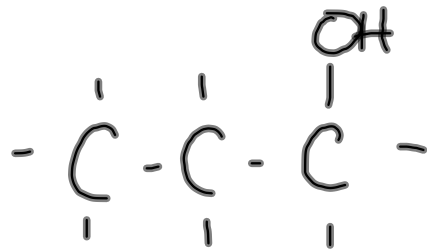
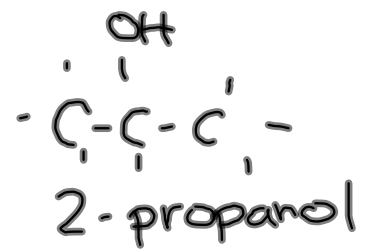
When there are multiple hydroxyl (-OH) groups, the alkane name is given, with the suffix indicating the number of -OH groups.

Ex.





2,2,3-butanetriol



1-propanol



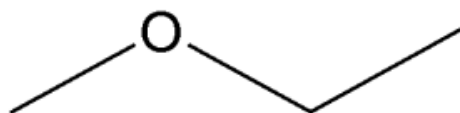
# Ethers

Ethers are organic molecules in which an oxygen is bonded to two carbon groups.



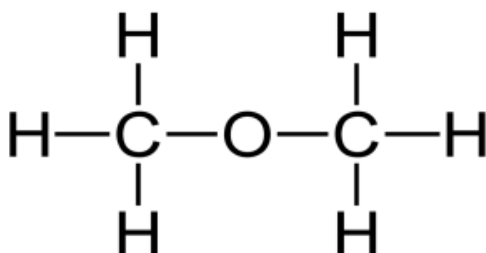
The alkyl groups attached to the oxygen atom are named in alphabetical order and are followed by the word *ether*.

Ex.



ethylmethyl ether

Ex.



dimethyl ether