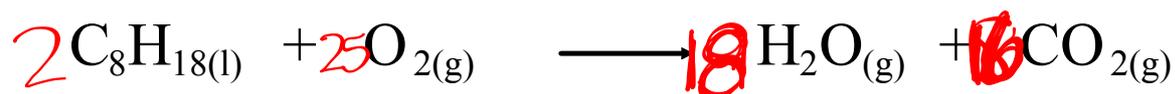
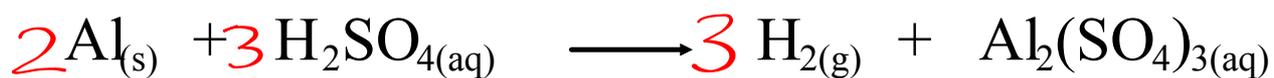


# Warm Up



## Check Homework - Worksheet

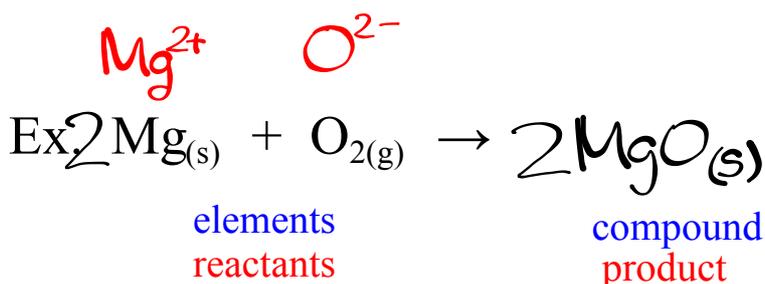


# Types of Chemical Reactions

There are five types of chemical reactions:

**I. Formation/Combination** reactions occur when two substances (normally elements) react to form an ionic or molecular compound

-when a metal and nonmetal react, the product will be the ionic compound formed by the most common ions.



**II. A decomposition** reaction is the result of an ionic or molecular compound breaking down into its elements.

⇒ it is the reverse of a formation reaction



compound  
reactant

elements  
products

Br HONClIF

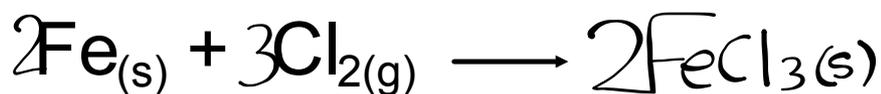
P<sub>4</sub> S<sub>8</sub>

DECOMP.

Sodium  
Carbonate

→ sodium + carbon + oxygen

FORMATION

 $\text{Fe}^{3+}$   $\text{Cl}^-$ 

iron + chlorine → iron(III)chloride

Copper (II) oxide →

\_\_\_\_\_

# Homework

## Worksheet