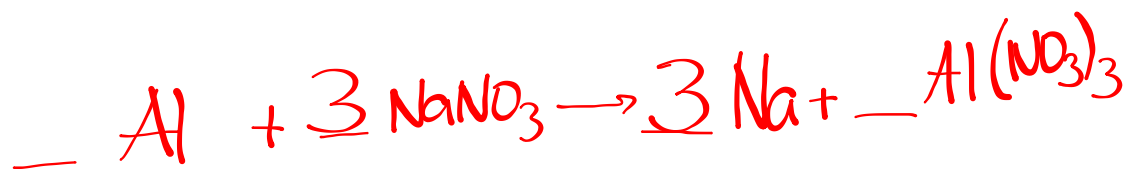
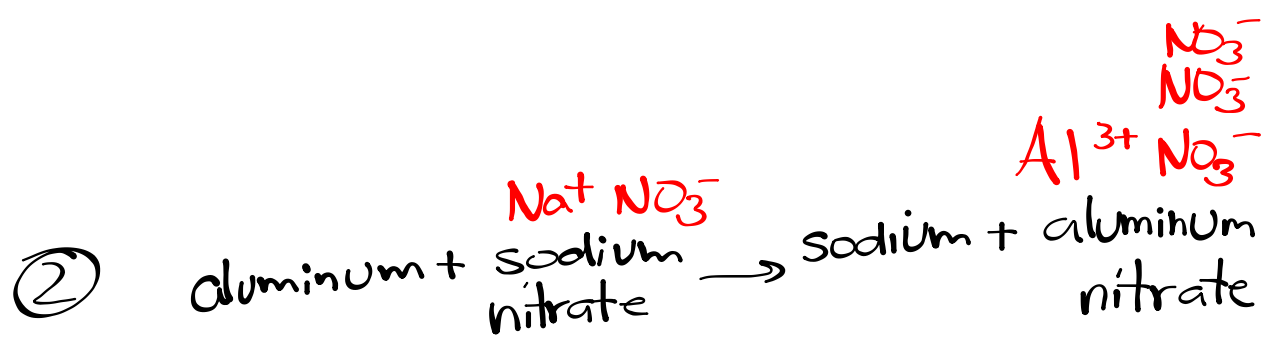


Check Homework



Reactions so far...

Combustion

element/compound + O₂ ⇒ oxides + energy



Synthesis

two smaller particles (elements) ⇒ one molecule



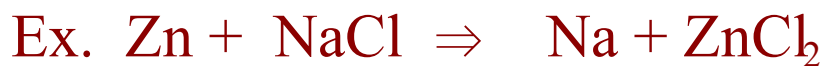
Decomposition

one molecule ⇒ smaller particles (elements)



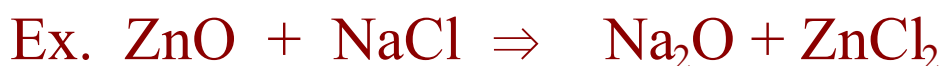
Single Replacement

element + compound ⇒ element + compound



Double Replacement

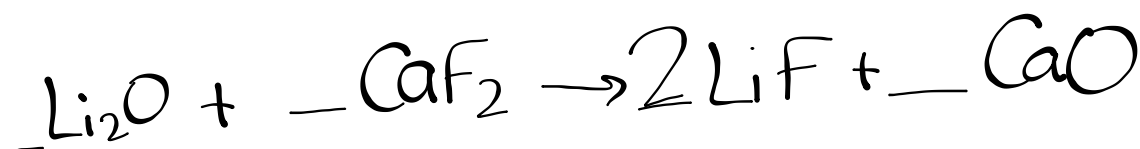
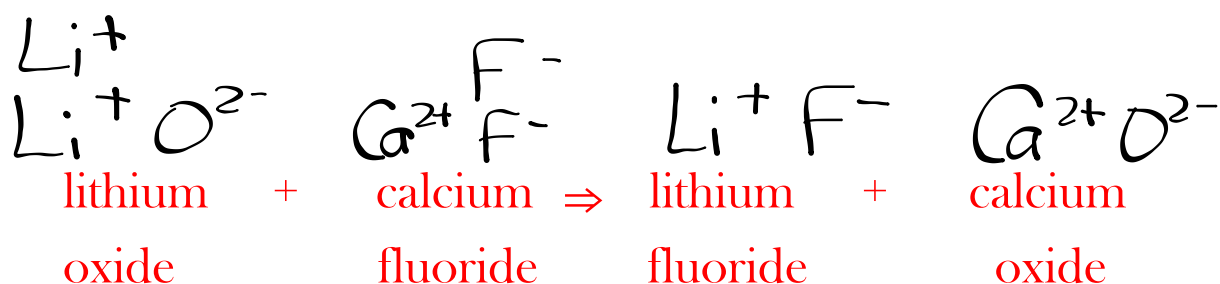
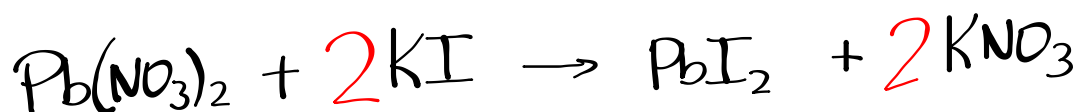
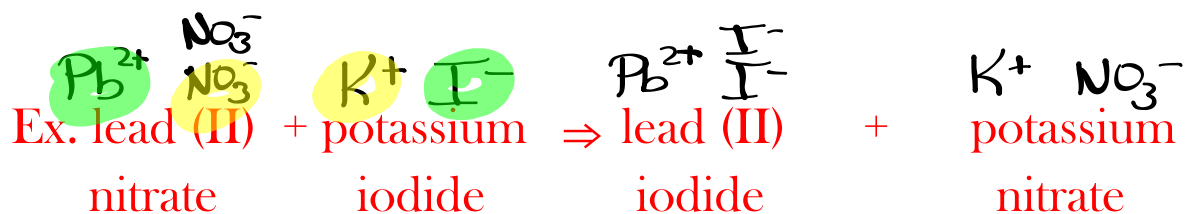
compound + compound ⇒ compound + compound



Double Replacement Reactions

Double replacement reactions are chemical changes that involve **two compounds** as reactants.

⇒ metals (or nonmetals) will 'trade'



Homework

p. 241 #1-3

