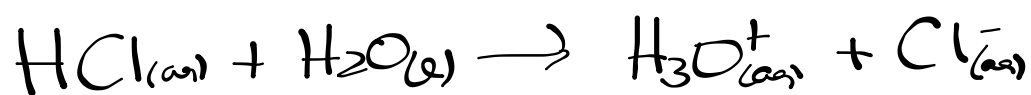
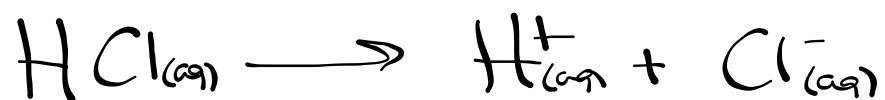


Acid-Base Chemistry Review

- Arrhenius Concept of Acid and Bases (Old and Revised)
- Bronsted-Lowry Acid-Base Concept
(Identify acids/bases, amphiprotic)
- Conjugate Acids and Bases
- Water Equilibrium
- Strong/weak Acids/Bases





↓
Ka =



↓
 K_b

$\text{pH} < 7$ ACIDIC

$\text{pH} = 7$ NEUTRAL

$\text{pH} > 7$ BASIC

