Warm Up

$$3Ca(NO_3)_{2(aq)} + 2AlCl_{3(aq)} \longrightarrow 3CaCl_{2(aq)} + 2Al(NO_3)_{3(aq)}$$

$$A|_{3+}$$
 $NO_3^ 2Al_{(s)} + 3Pb(NO_3)_{2(aq)} \longrightarrow 3Pb(s) + 2Al(NO_3)_{3(aq)}$ metal

Homework - #17-19

Chemical Reactions

A second type of double replacement reaction is aneutralization reaction, which is a reaction between an acid and a base, to form water and an ionic compound.

Worksheet