Warm Up

$$3^{\text{Fe}_{(s)}} + Al_2(SO_4)_{3(aq)} \longrightarrow 2^{\text{Al}_{(s)}} + 3^{\text{Fe}SO_{4(aq)}}$$

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Types of Chemical Reactions

There are five types of chemical reactions:

I. Formation/Combinationreactions occur when two substances (normally elements) react to form an ionic or molecular compound

-when a metal and nonmetal react, the product will be the ionic compound formed by the most common ions.

II. A decomposition reaction is the result of an ionic or molecular compound breaking down into its elements. ⇒it is the reverse of a formation reaction

$$\begin{array}{c} \text{Compound} \rightarrow \text{elements} \\ \text{Ex}_2 H_2 O_{(l)} \rightarrow 2 H_{2G} + O_{2G} \\ \text{compound} \\ \text{reactant} \end{array}$$
 elements products

$$K^{+}$$
 N^{3-}
 $6K_{(s)} + N_{2(g)} \longrightarrow 2K_{3}N_{(s)}$
Potassium + nitrogen \longrightarrow potassium nitride

Homework

Worksheet