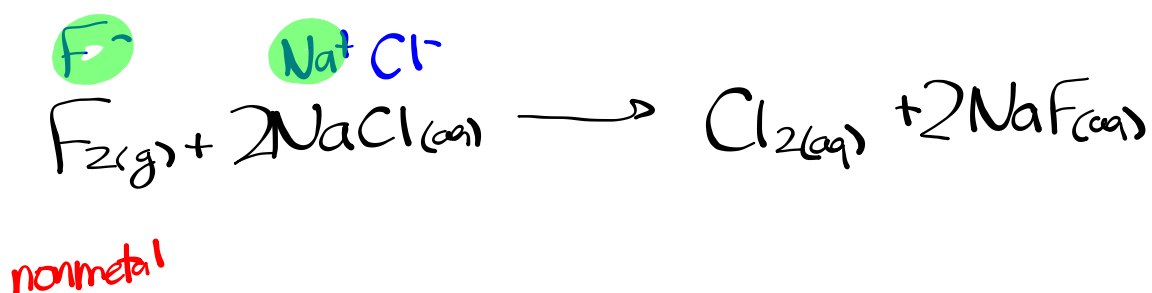
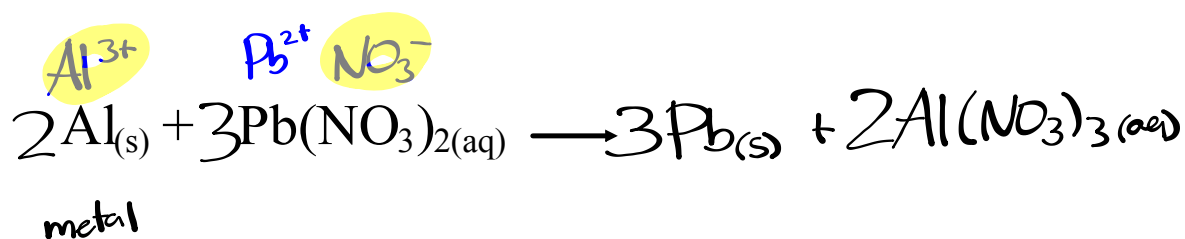
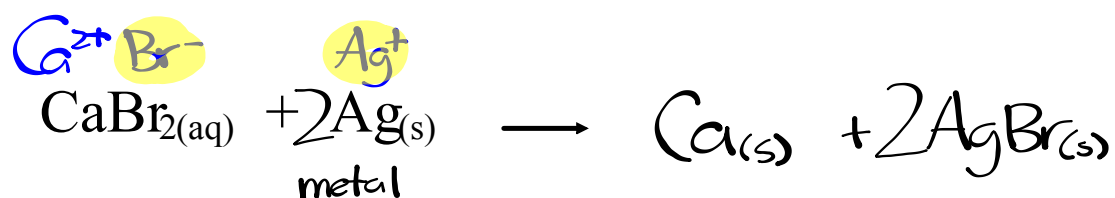


Warm Up

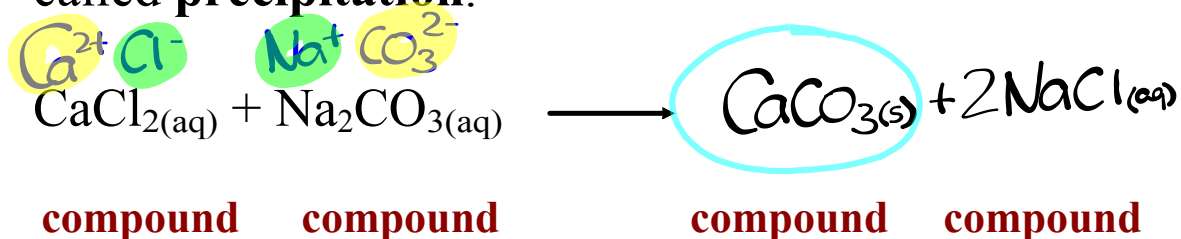


Chemical Reactions

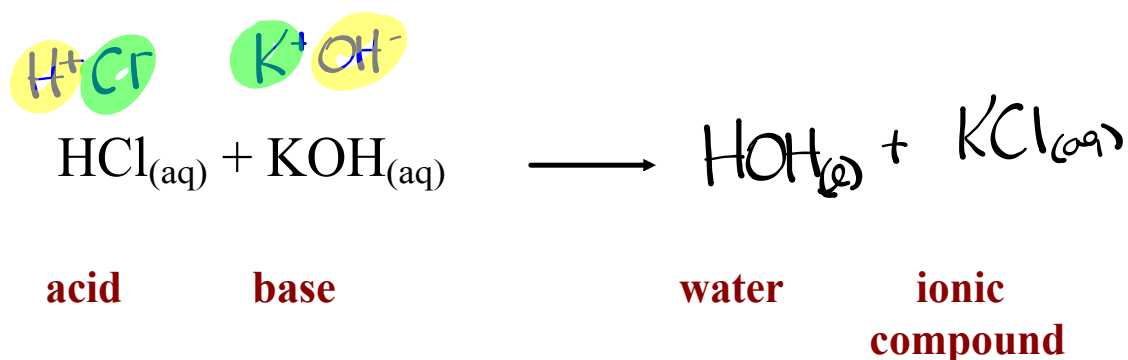
V. Double Replacement Reaction

Reaction that occurs between two ionic compounds in solution. Ions will "change partners".

⇒ if one of the products has low solubility, it may form a precipitate (solid). This double replacement reaction is called **precipitation**.



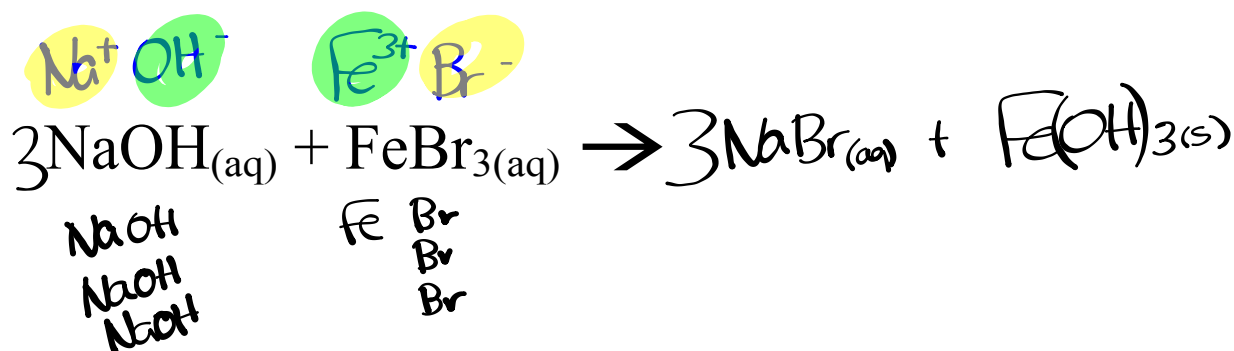
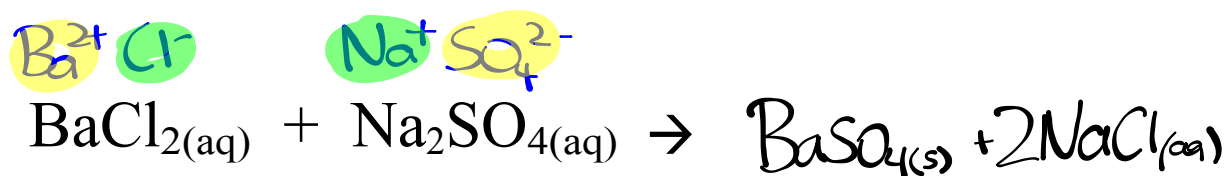
A second type of double replacement reaction is a **neutralization** reaction, which is a reaction between an acid and a base, to form water and an ionic compound.



DOUBLE REPLACEMENT

Compound + compound \rightarrow Compound + compound

Practice Problems



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