

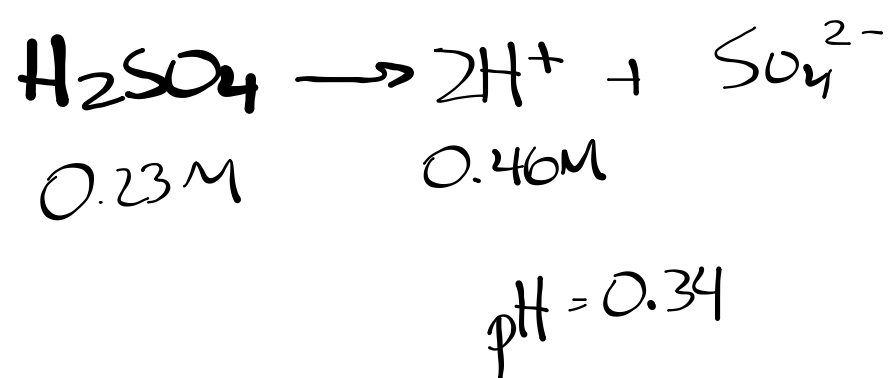
Organic Chemistry

Major Topics

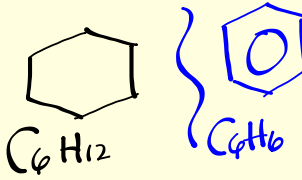
- Drawing / Naming Compounds
- Isomers
- Reactions

You Should Know...

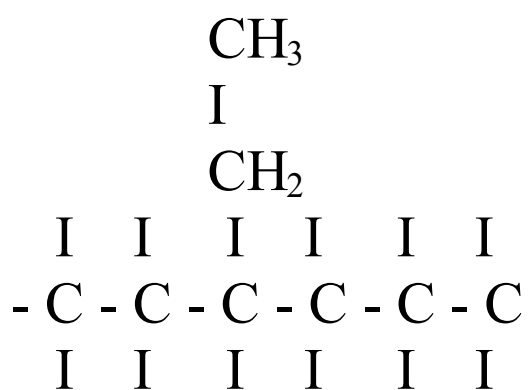
- General formulas of alkanes, alkenes, alkynes, and cyclic compounds
- Aromatic compounds
- Pi bonds
- Characteristics of organic compounds



Families of Organic Compounds (p.294)

Alkanes	ethane	$\begin{array}{c} \\ -\text{C} - \text{C}- \\ \end{array}$
Alkenes	ethene	$\begin{array}{c} \\ -\text{C} = \text{C}- \\ \end{array}$
Alkynes	ethyne	$-\text{C} \equiv \text{C}-$
Aromatics and Cyclic Compounds	cyclohexane	
Organic Halides	chloroethane	$\begin{array}{c} \\ -\text{C} - \text{C}-\text{Cl} \\ \end{array}$
Alcohols	ethanol	$\begin{array}{c} \\ -\text{C} - \text{C}- \\ \quad \\ \quad \text{OH} \end{array}$
Carboxylic Acids	ethanoic acid	$\begin{array}{c} \\ -\text{C} - \text{C} \begin{array}{l} \text{O} \\ \end{array} - \text{OH} \\ \end{array}$
Aldehydes	ethanal	$\begin{array}{c} \text{O} \\ \\ -\text{C} - \text{C}- \\ \end{array}$
Ketones	propanone	$\begin{array}{c} \text{O} \\ \\ -\text{C} - \text{C} - \text{C}- \\ \quad \end{array}$
Esters	methyl ethanoate	$\begin{array}{c} \text{O} \\ \\ -\text{C} - \text{C} - \text{O} - \text{C}- \\ \quad \end{array}$
Ethers	ethylmethyl ether	$\begin{array}{c} \\ -\text{C} - \text{O} - \text{C} - \text{C}- \\ \quad \end{array}$

Draw ethyl-2,4-dimethylpentane



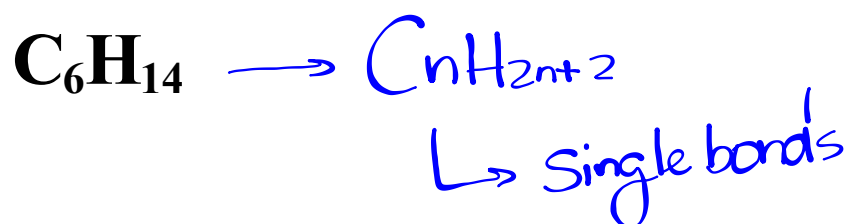
Naming Hints:

- List branches alphabetically
- Location of multiple bond takes precedence over branch
- Esters - name alcohol part, then carboxylic acid part

Ex. ethanoic acid + methanol --->

Isomers

Name and draw all isomers for the following:



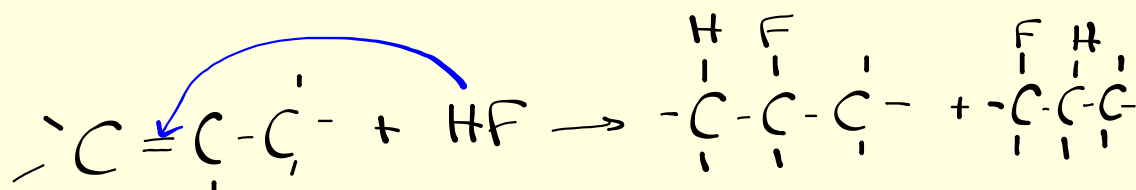
Name and draw all isomers for the following:



Reactions

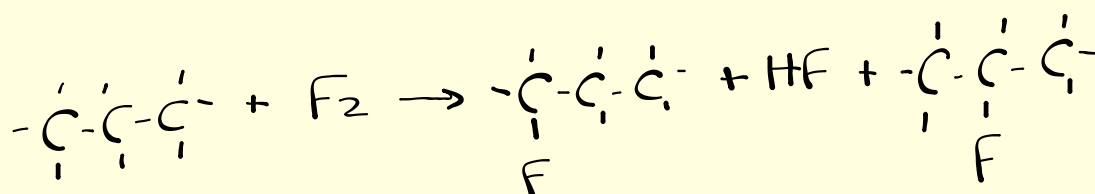
Addition

alkene/alkyne + H₂ or HX or X₂



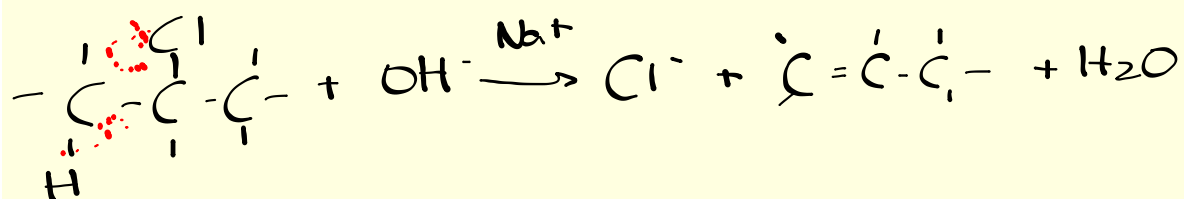
Substitution

alkane/aromatic + halogen



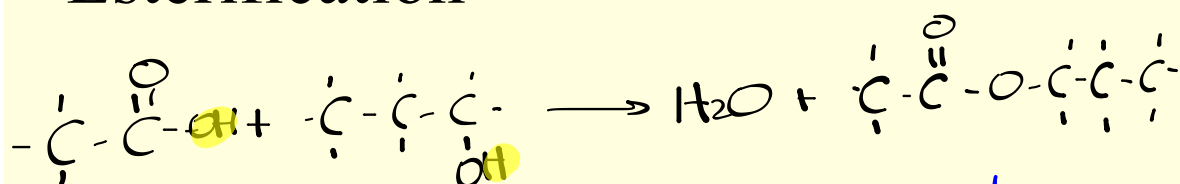
Elimination

alkyl halide + OH⁻
alcohol + acid



Esterification

carboxylic acid + alcohol



ethanoic acid + 1-propanol → water + propyl ethanoate

Cracking

Formation