## Pass in Assignment

## Worksheet #2

Agrin + Sgran Agr Sgran Agr Sgran Mon-metal silver sulfide 
$$G^{3+}G^{3+}S^{2-$$

## **Writing Formulas for Polyatomic Ions**

polyatomic ions- are ions made up of more than one atom.

Determining formulas of polyatomic ion compounds are done in the same manner as binary compounds.

Ex. sodium carbonate

Nat 
$$CO_3^{2-}$$
Nat  $CO_3^{2-}$ 

Ex. calcium nitrate

$$G_{3}^{2+} NO_{3}^{-}$$
  $G_{3}^{-} NO_{3}^{-}$ 



## **Naming Polyatomic Ionic Compounds**

Ex. CaCO<sub>3</sub>

Calcium Carbonate

Name the following polyatomic compounds:

(b) 
$$Fe_2(CO_3)_3$$
  
 $Fe^{3+}$   $CO_3^{2-}$   
 $Fe^{3+}$   $CO_3^{2-}$   
 $CO_3^{2-}$ 

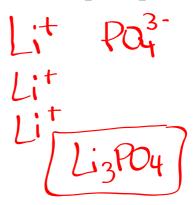
Write the chemical formula for the following polyatomic compounds:

(c) calcium hydroxide

$$G^{2+} OH^{2}$$

$$OH^{2}$$

(d) lithium phosphate



Sheet - Ionic Compounds

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