Quiz

Lonic · metal /nonmetal - attraction (Ionic bord) Nacı Nat CI Sodium Chloride

Molecular

nonmetal/honmeta

want

Share electrons

(cavalent bond)

NO2

NO2

Molecular Compounds

MOLECULAR THEORY -nonmetal atoms share electrons in a covalent bond to attain a maximum number of valence electrons (complete outer shell) rather than gaining electrons from metal atoms.

Ex. CO₂

<u>Molecular elements</u>- although the chemical formula of metals are frequently shown alone as a single atom (Na), nonmetals frequently form **diatomic molecules**.

Ex. H₂, N₂, O₂, F₂, Cl₂, Br₂, I₂

Also: O₃, P₄, S₈

Naming Binary molecular compounds

As outlined by IUPAC rules, some molecular compounds signify the number of atoms in the molecular formula by using the same prefixes as hydrates.

Ex. CS₂

Carbon disulfide

see Table 9.4 p. 269

The prefix system is usually not used for hydrogen molecular compounds

Ex. water - H_2O_2

N205

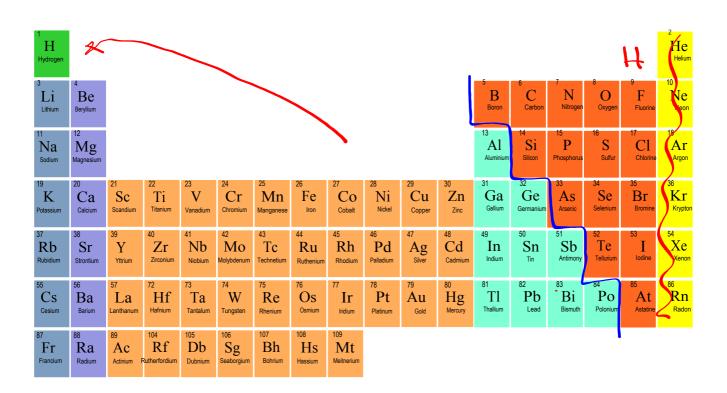
dinitrogen pentoxide

N20

dinitrogen monoxide

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Worksheets



| Cerium P | Pr Praseodymium | Nd Neodymium | Promethium | Samarium | Europium | Gadolinium | Tb | Dy Dysprosium | Ho Holmium | Erbium | Tm | Yterbium | Lu Lutetium |
|------------------|--------------------|--------------------|-----------------------|--------------|-----------------------|--------------------|------------------------------|-------------------------|-------------------------|---------|--------------------------|-------------|-------------------------|
| 90 Th Thorium | Protactinium | 92 U Uranium | 93 Np Neptunium | Pu Plutonium | 95 Am Americium | 96 Cm Curium | 97 Bk Berkelium | 98 Cf Californium | 99 Es Einsteinium | Fermium | 101 Md Mendelevium | No Nobelium | 103 Lr Lawrencium |