In-Class Assignment??

Homework - Worksheet #1

- (Da) HgS
 - b) PbS2

 - c) ZnO d) MnF4

- e) G_3P_2
- f) NiBr2
- g) CuCl₂ h) FeI₃

Worksheet #2

Writing Formulas for Polyatomic Compounds

polyatomic ions- are ions made up of more than one atom.

Ex. CO₃²- carbonate ion

All polyatomic ions can be found with their charges on the back cover of the periodic table.

Determining formulas of polyatomic ion compounds are done in the same manner as binary compounds.

Ex. sodium carbonate

$$Na^{t}$$
 Co_3^{2-}

162CO3

Ex. calcium nitrate

$$G^{2+} NO_3^{-}$$







ammonium

Naming Polyatomic Ionic Compounds

Ex. CaCO₃

Calcium carbonate

Name the following polyatomic compounds:

(a) NaHCO₃

No+ HCO3-

Sodium bicarbonate

(b) $Fe_2(CO_3)_3$

Fe3+ CO3- CO3-

iron (III) carbonate

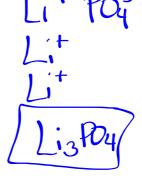
Write the chemical formula for the following polyatomic compounds:

(c) calcium hydroxide

(2+ OH)

(a(OH)2

(d) lithium phosphate



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Sheet - Ionic Compounds