

Ionic and Molecular Compounds Review Worksheet

Write the name for each of the following Ionic Compounds:

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|--------------------------------------|-----------------------|---------------------------------------|------------------------|
| 1. NaBr | sodium bromide | 6. VCl ₄ | vanadium (IV) chloride |
| 2. Ti(SO ₄) ₂ | titanium (IV) sulfate | 7. Mg ₃ P ₂ | magnesium phosphide |
| 3. FePO ₄ | iron (III) phosphate | 8. Sr(HCO ₃) ₂ | strontium bicarbonate |
| 4. K ₃ N | potassium nitride | 9. Ba ₃ N ₂ | barium nitride |
| 5. CuOH | copper (I) hydroxide | 10. Zn(NO ₃) ₂ | zinc nitrate |

Write the name for each of the following Molecular Compounds:

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|----------------------------------|-------------------------------|-----------------------------------|-----------------------|
| 1. CO | carbon monoxide | 6. SO ₂ | sulfur dioxide |
| 2. P ₂ O ₅ | diphosphorous pentaoxide | 7. N ₂ O | dinitrogen monoxide |
| 3. CO ₂ | carbon dioxide | 8. N ₂ O ₄ | dinitrogen tetraoxide |
| 4. CH ₄ | carbon tetrahydride / methane | 9. BCl ₃ | boron trichloride |
| 5. CF ₄ | carbon tetraflouride | 10. C ₂ S ₄ | dicarbon tetrasulfide |

Write the formula for each of the following Ionic Compounds:

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|-----------------------------|------------------------|------------------------|---------------------|
| 1. nickel (III) sulfide | <chem>Ni2S3</chem> | 6. potassium carbonate | <chem>K2CO3</chem> |
| 2. copper (II) sulfate | <chem>CuSO4</chem> | 7. silver oxide | <chem>Ag2O</chem> |
| 3. manganese (II) phosphate | <chem>Mn3(PO4)2</chem> | 8. tin (IV) selenide | <chem>Sn3Se4</chem> |
| 4. aluminum phosphate | <chem>AlPO4</chem> | 9. lithium nitride | <chem>Li3N</chem> |
| 5. magnesium hydroxide | <chem>Mg(OH)2</chem> | 10. copper (I) sulfide | <chem>Cu2S</chem> |

Write the formula for each of the following Molecular Compounds:

- | | | | |
|-------------------------|--------------------|--------------------------|-------------------|
| 1. silicon dioxide | <chem>SiO2</chem> | 6. hydrogen peroxide | <chem>H2O2</chem> |
| 2. diboron tetrabromide | <chem>B2Br4</chem> | 7. dicarbon tetrasulfide | <chem>C2S4</chem> |
| 3. carbon tetrachloride | <chem>CCl4</chem> | 8. nitrogen tribromide | <chem>NBr3</chem> |
| 4. nitrogen monoxide | <chem>NO</chem> | 9. methane | <chem>CH4</chem> |
| 5. sulfur dioxide | <chem>SO2</chem> | 10. oxygen | <chem>O2</chem> |