## Science 10 Chapter 6 Chemical Reactions Test Review

Part A: Make sure you can match the following terms to their definitions

Coefficient Skeletal chemical equation Double replacement

Incomplete Subscript Reactants

combustion Synthesis Law of conservation of

Decomposition Products mass

Part B: Complete questions pg 252-253 #5, 6, 12

Part C: For each of the following combustion reactions, complete the balanced chemical reaction

- a)  $C_5H_{12} + O_2 \rightarrow$  complete combustion
- b)  $C_4H_{10} + O_2 \rightarrow$  incomplete combustion
- c)  $C_6H_{14} + O_2 \rightarrow$  complete combustion

Part D: Solution A has a mass of 103g. Solution B has a mass of 55g. When they are mixed, a chemical reaction occurs in which a gas is produced. If the mass of the final mixture is 155g, what mass of gas was produced? Define the law of conservation of mass using this example.