

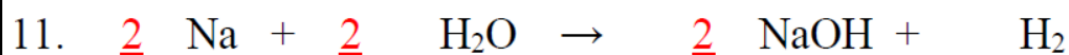
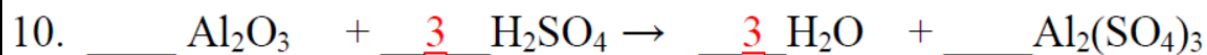
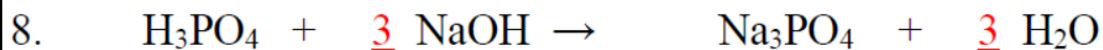
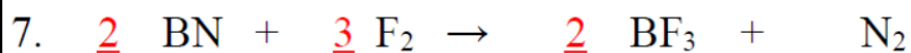
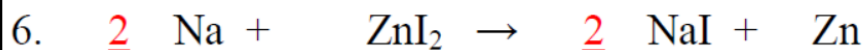
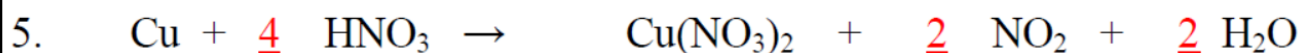
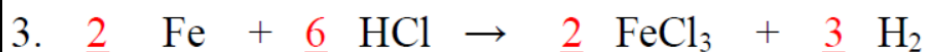
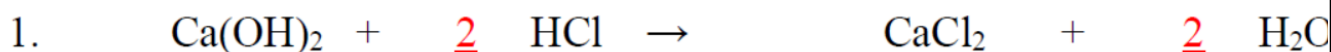
Oct 9, 2019

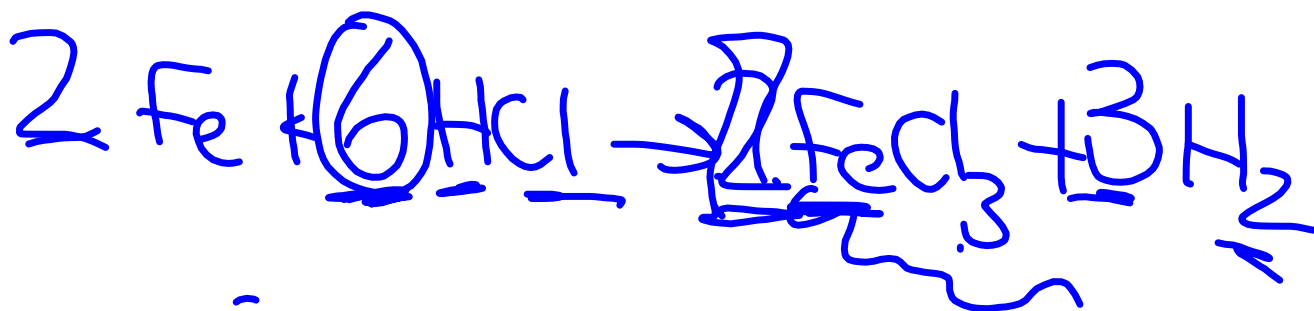
answers balancing equations WS
more practice balancing equations

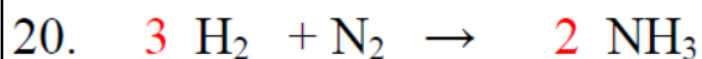
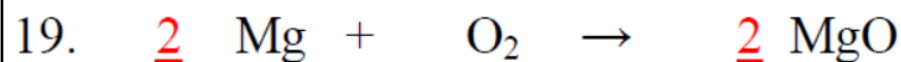
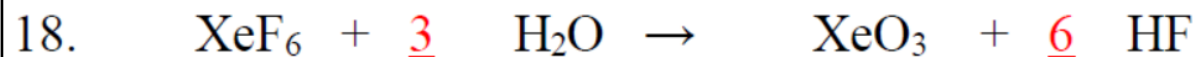
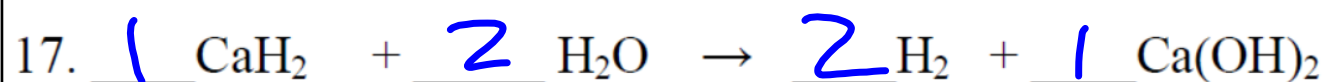
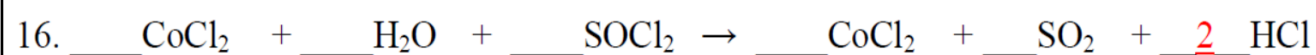
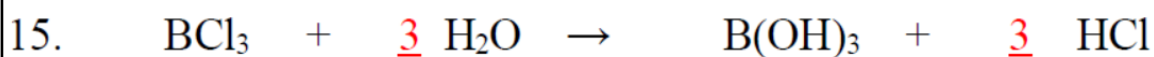
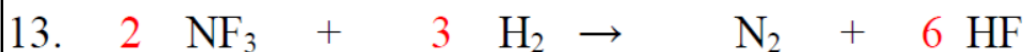
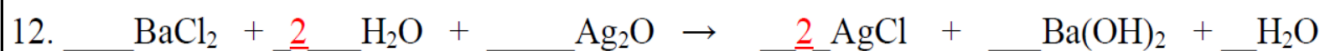
**Quiz on Balancing
Equations Tomorrow!!!**

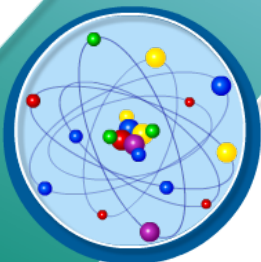
Physical Science 10
Balancing Equations

Balance each of the following equations. Once you have balanced the equations name each of the chemical formulas involved in the reaction.



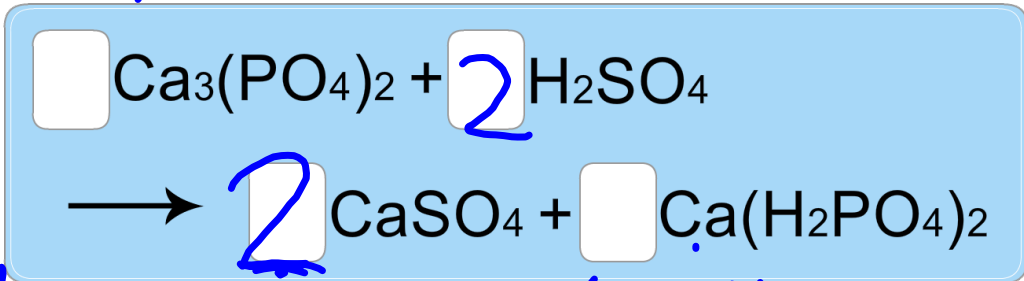
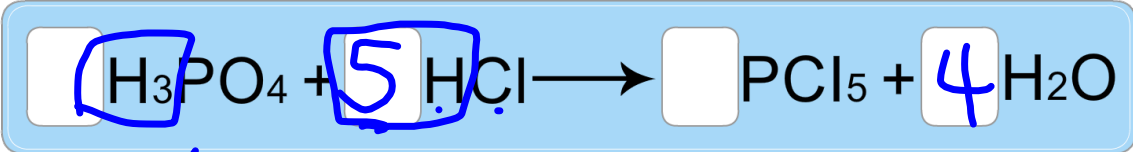
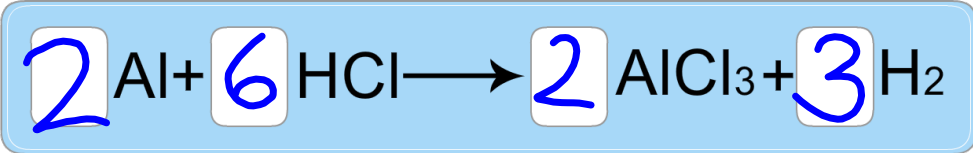






Practice Balancing Equations

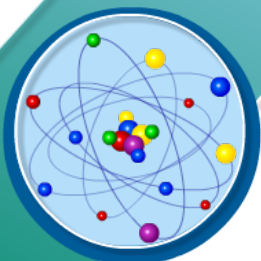
Write the correct coefficient next to each molecule to balance the equation.



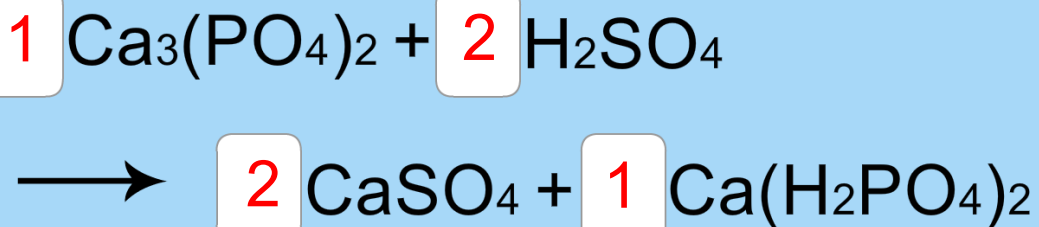
Check Answer

Al	2	2
H	6	6
Cl	6	6
P	1	1
O	4	4
Ca	3	3
PO ₄	2	2
H	4	4
SO ₄	2	2

Al	2	2
H	6	6
Cl	6	6
P	1	1
O	4	4
Ca	3	3
PO ₄	2	2
H	4	4
SO ₄	2	2



Correct Answer



Correct Answer

Extra Practice Balancing Equations WS's

- * Remember you can only add co-efficient's
- * Balance the easy atoms first (those that only appear once on each side of the equation)
- * Keep poly-atomics that stay together together.
- * Keep oxygen till the end (if there is an odd number of oxygen after you have balanced everything else put the odd number in front of the O_2 that is alone and double the remaining numbers)
- * If you have an OH on one side and an H_2O on the other you can re-write the H_2O as HOH and keep the polyatomic together.

Quiz Tomorrow! Make sure you ask questions if you do not understand!!!